Macromolecules

Volume 34, Number 17

August 14, 2001

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Communications to the Editor

Slip, Concentration Fluctuations, and Flow Instability in Sheared Polymer Solutions

William B. Black

3M Corporate Process Technology Center, St. Paul, Minnesota 55144

Michael D. Graham*

Department of Chemical Engineering and Rheology Research Center, University of Wisconsin—Madison, Madison, Wisconsin 53706-1691

Received April 30, 2001 Revised Manuscript Received June 14, 2001

The local composition of polymer solutions and blends can be significantly affected by flow. For example, shearenhanced concentration fluctuations in semidilute polystyrene (PS) solutions have been observed via lightscattering and microscopy by a number of groups. 1-8 A purely hydrodynamic mechanism for this phenomenon, in an unbounded flow domain, was proposed initially by Helfand and Fredrickson (HF)9 and later extended by several authors. 10-14 This mechanism is based on the idea that variations in polymer stress can drive a flux of polymer molecules—indeed, several authors^{12,13,15–17} have developed two-fluid models which demonstrate that the polymer flux j in a solution has the form j = $-D\nabla n + (D/k_{\rm B}T)\nabla \cdot \boldsymbol{\tau}$, where *n* is the polymer number density, τ is the polymer extra stress tensor, D is the polymer gradient (collective) diffusivity, $k_{\rm B}$ is the Boltzmann constant, and T is the temperature. A scattering peak is predicted and observed to arise when the diffusion and relaxation times are equal, which occurs at a wavelength of roughly $\sqrt{D\lambda}$, where λ is the longest relaxation time for the solution. For a semidilute PS solution, we note that this scale can be several micrometers, much larger than the equilibrium molecular size. 18

A recent experimental study by Mhetar and Archer¹⁹ has demonstrated several novel features of shearenhanced concentration fluctuations in entangled polymer solutions. They studied semidilute solutions of PS in diethylphthalate (DEP), to which tracer particles were added to measure the velocity profile, using a planar Couette cell with (high energy) silica walls. In this situation, wall-slip occurs primarily due to the disentanglement of bulk chains from those adsorbed to the walls, rather than to the formation of a depletion layer, as arises in dilute solutions.^{20,21} Three features of the flow of these solutions stand out. First, no turbidity was observed at low shear rates $(\bar{\gamma})$ where slip was weak; enhanced concentration fluctuations were observed, however, at large shear rates (Weissenberg number $We = \lambda \bar{\dot{\gamma}} \gg 1$), where slip was significant. The length scale for the fluctuations was observed to be $O(10 \, \mu \text{m})$, as was the extrapolation, or slip, length b measured for these solutions. Second, the concentration fluctuation enhancement began near the boundaries, and third, modification of the surface to suppress adsorption, thereby changing the mechanism of slip from disentanglement to "bare" friction between polymer and substrate, delayed the onset on fluctuations to much larger shear stresses. These observations strongly suggest that slip plays a role in the formation and development of these fluctuations. Our goal here is to analyze the coupling between concentration, stress, and slip in bounded shear flow, to elucidate the effects of boundaries on the dynamics and to attempt to capture in a simple model the essential phenomena observed by Mhetar and Archer.

We consider plane shear flow, as shown in Figure 1. To describe the behavior of the solution, we use the two-fluid model derived by Mavrantzas and Beris, 16 simplified to a solution of Hookean dumbbells, 26 and written in terms of the conformation (second moment) tensor, α . Nondimensionalizing stresses with n_0k_BT , where n_0 is the average concentration of the solution, lengths with the characteristic length $\sqrt{D\lambda}$, velocities with $\bar{\gamma}\sqrt{D\lambda}$, time with λ , and concentration with n_0 gives, to leading

^{*} Corresponding author. E-mail: graham@engr.wisc.edu.

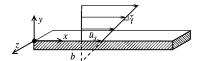


Figure 1. Plane Couette geometry showing slip between the solution and solid surface. The flow is in the x direction, y is the gradient direction, and z is the vorticity or neutral direction. The steady state slip velocity is \bar{u}_s , $\bar{\gamma} = \partial \bar{v}_s / \partial y$ is the steady state shear rate, and $b \equiv \bar{u}_s / \bar{\gamma}$ is the extrapolation length.

order when polymer volume fraction $\phi \ll$ 1, the following model equations

$$\frac{\partial \mathbf{\alpha}}{\partial t} + We \left(\mathbf{v} \cdot \nabla \mathbf{\alpha} - \mathbf{\alpha} \cdot \nabla \mathbf{v}_{p} - (\nabla \mathbf{v}_{p})^{T} \cdot \mathbf{\alpha} \right) + \mathbf{\alpha} - n \boldsymbol{\delta} = \nabla^{2} \mathbf{\alpha}$$
(1)

$$\frac{\partial n}{\partial t} + We \mathbf{v} \cdot \nabla n = -\nabla \cdot (-\nabla n + \nabla \cdot \mathbf{r})$$
 (2)

$$-\nabla p + \nabla \cdot \boldsymbol{\tau} + SWe\nabla^2 \mathbf{v} = 0 \tag{3}$$

$$\nabla \cdot \mathbf{v} = 0 \tag{4}$$

where \mathbf{v} is the mass-average velocity, $\mathbf{v}_{p} = \mathbf{v} + (-\nabla n +$ $\nabla \cdot \boldsymbol{\tau}$) is the polymer velocity (polymer flux/polymer concentration), S is the ratio of the solvent to polymer viscosities, *n* is the number density of dumbbells, $\tau = \alpha$ $-n\delta$ is the polymer extra stress and p is the pressure. (Here and hereafter, all references to variables refer to their nondimensional form). Equation 1 describes the evolution of polymer conformation. Equation 2 is the polymer species conservation equation. If diffusion of the polymer is neglected, the right-hand sides of eqs 1 and 2 vanish, yielding the Oldroyd-B model. Equations 3 and 4 describe conservation of momentum and mass in the low Reynolds number limit. In presenting the model, we have suppressed any dependencies of properties on concentration—in the analysis below we only consider small perturbations from spatially uniform stress and concentration fields, in which case the terms that reflect these dependencies do not appear. Other phenomenologically motivated models for such systems are virtually identical to this one. 9,10,12-14 Previously, however, only Sun et al.¹⁴ have included the diffusive term in eq 1 in a treatment of scattering.

In Mhetar and Archer's experiments, either high energy (strongly adsorbing silica) or very low-energy (nonadsorbing octadecyltrichlorosilane-coated silica) surfaces were used. In either case, a reasonable first approximation is that in the course of an experiment, adsorbed chains do not desorb to a significant degree in flow, and that no additional bulk chains adsorb. Therefore, we apply no-flux boundary conditions for the polymer and the solvent: $\mathbf{n} \cdot \mathbf{v}_p = \mathbf{n} \cdot \mathbf{v} = 0$, where \mathbf{n} is the unit normal on the boundary. For the same reason, we assume that there are no sources or sinks for polymer conformation at the walls, giving a no-flux boundary condition for the conformation: $\partial \alpha/\partial y = 0$.

The model is completed by specifying the slip boundary condition. We use two simple models here. The first, more suited to the high energy surface, is a simple Navier slip boundary condition $u_s = b\tau_{yx}/We$, with the extrapolation length b taken to be a constant. This model crudely captures the fact that slip is related to polymer stress for the high energy surface (i.e., interaction between adsorbed and bulk chains). The other is a

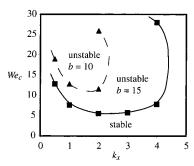


Figure 2. Neutral curves for $k_z = 0$, S = 0.01, with the stress-dependent slip model and various values of the extrapolation length. Key: (\blacktriangle) b = 10; (\blacksquare) b = 15. The curves are to guide the eye.

quasisteady relation $u_s = b\gamma$, again with b constant. This is the "Newtonian" limit of the Navier model and is appropriate if the slip is due to monomer friction, which will have an extremely short relaxation time, yielding an instantaneous coupling between slip and shear rate.

We have studied the dynamics of this model by linear stability analysis of the plane shear base state, given (for either slip model) $\mathbf{a} = (\bar{v}_x, \bar{v}_y, \bar{v}_z, \bar{\tau}_{xx}, \bar{\tau}_{yx}, \bar{\tau}_{zx}, \bar{\tau}_{yy}, \bar{\tau}_{zy}, \bar{\tau}_{zy}$

If the possibility of concentration fluctuations is suppressed in the model (by setting *D* to zero, thus eliminating the rhs of eqs 1 and 2), plane shear flow with no-slip²² or either of the above forms of Navier $slip^{24,25,29}$ is stable for all Weissenberg numbers. In addition, if concentration variations are permitted but slip is not (b = 0), the flow is also stable. In contrast, we find that if Navier slip and concentration variations are allowed, the flow can become unstable when We exceeds a critical value Wec, as observed in the experiments of Mhetar and Archer. 19 Two-dimensional disturbances ($k_z = 0$) are the most unstable, in agreement with experiment; Figure 2 shows a neutral curve plotted as We_c vs k_x for $k_z = 0$, with b = 10 and b = 15—these values of b are consistent with the slip lengths reported by Mhetar and Archer, assuming that $\sqrt{D\lambda}$ is several micrometers. ¹⁻⁸ For a particular wavenumber k_x , if Weis in the "unstable" region on the plot for a given b, then infinitesimal perturbations of that wavenumber grow. From these results, the critical length scale for the instability is predicted to be $O(\sqrt{D\lambda})$, and it arises when the slip length $b > \sqrt{D\lambda}$. In particular, we find that the dimensionless group $bk_x \approx 20-30$ at criticality. Experimentally, approximating k_x as $2\pi/(10 \mu m)$ and bas 40 μ m yields $bk_x \approx 25$. The neutral curves turn back on themselves as k_x increases, so sufficiently short wavelength disturbances always decay. Figure 3 shows a gray scale plot of a typical destabilizing concentration perturbation, which is clearly localized near the solid surface, as also seen in the experiments. Finally, if we



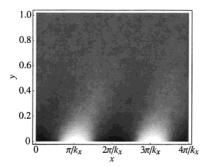


Figure 3. Concentration field of the destabilizing disturbance. $k_x = 2.0$, b = 15, We = 6, and S = 0.01. The pattern travels to the right with speed $c \approx \bar{u}_s$. Black indicates a lower concentration than the average, and white, higher.

replace the Navier boundary condition with its instantaneous version corresponding to monomer friction, we observe no instability, again consistent with the experimental observations.

To summarize, these theoretical results corroborate the experimental evidence presented by Mhetar and Archer¹⁹ that the nature of the bounding surfaces plays an important role in the formation of enhanced concentration fluctuations in sheared polymer solutions.

Acknowledgment. The authors gratefully acknowledge the National Science Foundation and the nontenured faculty grant program of the 3M Co. for supporting this work. We are also grateful to Prof. R. G. Larson for enlightening discussions.

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MA0107455